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Key indicators

Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.055
 wR factor = 0.165
Data-to-parameter ratio = 13.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**(E)-4-[4-(4-Chlorobenzoyloxy)-3-ethoxy-
benzylideneamino]-1,5-dimethyl-2-phenyl-
1H-pyrazol-3(2H)-one**

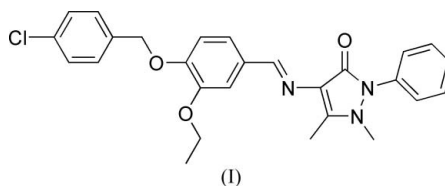
The title compound, $\text{C}_{27}\text{H}_{26}\text{ClN}_3\text{O}_3$, was prepared by reacting 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one and 4-(4-chlorobenzoyloxy)-3-ethoxybenzaldehyde. The central ethylvanillin group makes dihedral angles of 63.00 (11), 31.83 (7) and 74.47 (8)° with the chlorobenzene ring, the pyrazolone ring and the terminal phenyl ring, respectively. Packing is stabilized by intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds that form centrosymmetric dimers.

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Comment

Schiff base ligands have received a good deal of attention in the development of coordination chemistry for more than 100 years (Kahwa *et al.*, 1986). Among the large number of compounds, 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one forms a variety of Schiff bases with aldehydes (Han & Zhen, 2005; Shi, 2005). In the present study we report the synthesis and molecular structure of such a compound, (I).



In the title molecule (Fig. 1), bond lengths and angles are within normal ranges. The central ethylvanillin group (C8–C13/C16/O1/O2) is planar, with an r.m.s. deviation for fitted atoms of 0.0224 Å, and it makes dihedral angles of 63.00 (11), 31.83 (7) and 74.47 (8)° with the benzene ring (C1–C6), the pyrazolone ring (C17/C18/N2/N3/C21/N1/O3) and the terminal phenyl ring (C22–C27), respectively. The pyrazolone ring is also almost planar, with an r.m.s. deviation for fitted atoms of 0.0370 Å. It makes a dihedral angle of 48.41 (9)° with the terminal phenyl ring. The crystal packing is stabilized by weak non-classical intermolecular $\text{C}-\text{H}\cdots\text{O}=\text{C}$ hydrogen bonds (Table 1) that form centrosymmetric dimers (Fig. 2).

Experimental

An anhydrous (99.5%) ethanol solution of 4-(4-chlorobenzoyloxy)-3-ethoxybenzaldehyde (2.91 g, 10 mmol) was added to an anhydrous ethanol solution of 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one (2.03 g, 10 mmol) and the mixture stirred at 350 K for 3 h under nitrogen, whereupon a yellow precipitate appeared. The product was then isolated and recrystallized from ethanol, and dried *in vacuo* to give pure (I) in 82% yield. Yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution.

Crystal data

 $C_{27}H_{26}ClN_3O_3$ $M_r = 475.96$ Triclinic, $P\bar{1}$ $a = 10.019 (2) \text{ \AA}$ $b = 10.355 (2) \text{ \AA}$ $c = 12.824 (3) \text{ \AA}$ $\alpha = 77.080 (4)^\circ$ $\beta = 83.710 (4)^\circ$ $\gamma = 71.133 (4)^\circ$ $V = 1226.1 (4) \text{ \AA}^3$ $Z = 2$ $D_x = 1.289 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.19 \text{ mm}^{-1}$ $T = 294 (2) \text{ K}$

Block, yellow

 $0.34 \times 0.30 \times 0.20 \text{ mm}$

Data collection

Bruker SMART APEX CCD area-detector diffractometer

6294 measured reflections

4299 independent reflections

 φ and ω scans2253 reflections with $I > 2\sigma(I)$

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

 $R_{\text{int}} = 0.030$ $T_{\text{min}} = 0.928, T_{\text{max}} = 0.963$ $\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.055$ $wR(F^2) = 0.165$ $S = 1.05$

4299 reflections

310 parameters

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0704P)^2 + 0.1725P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.003$ $\Delta\rho_{\text{max}} = 0.41 \text{ e \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry ($\text{\AA}, ^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C26-H26\cdots O3^i$	0.93	2.60	3.395 (4)	144

Symmetry code: (i) $-x, -y - 1, -z + 1$.

All H atoms were included in calculated positions and refined using a riding-model. Constrained C—H bond lengths and isotropic $U_{\text{iso}}(\text{H})$ parameters: 0.93 \AA and $1.2U_{\text{eq}}(\text{C})$ for aromatic, 0.97 \AA and $1.2U_{\text{eq}}(\text{C})$ for methylene, and 0.96 \AA and $1.5U_{\text{eq}}(\text{C})$ for methyl H atoms.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 1997); software used to prepare material for publication: SHELXL97.

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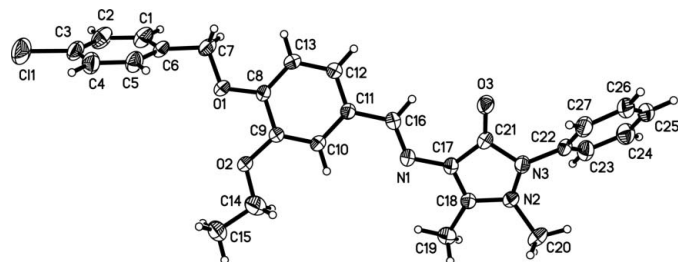


Figure 1

The structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

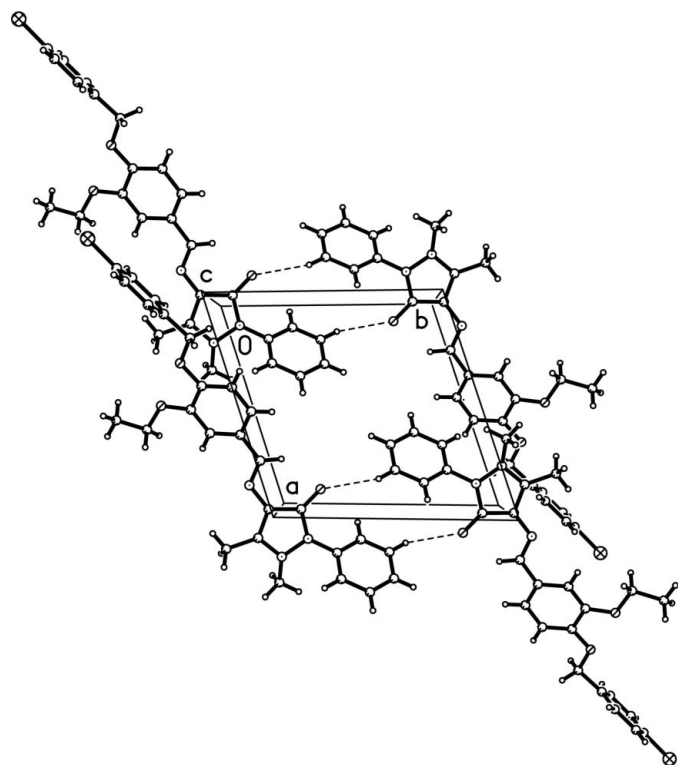


Figure 2

A view, down the c axis, of the packing arrangement in the crystal structure of (I). Intermolecular hydrogen bonds are represented by dashed lines.

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